

N-COORDINATED TETRYLENES

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Tetrylenes of composition $[L_2E:]$ or $[L(X)E:]$ ($E = Si, Ge, Sn, Pb$; $L =$ supporting ligand; $X =$ polar group) attract attention in the present time due to their unique structures and reactivities.¹ As the central atom E has both pair of electron and vacant p orbital, tetrylenes may behave as donors and acceptors. While electrophilic tetrylenes are capable of activating small molecules, tetrylenes supported by amidinate, β -diketiminato and other chelate ligands are good donating ligands and may coordinate transition or main group metals.³ Some of these complexes are able to catalyse organic transformations.³ Tetrylenes containing different type of chelating ligands are also studied in our department. Therefore, chemistry of N -coordinated tetrylenes will be presented.

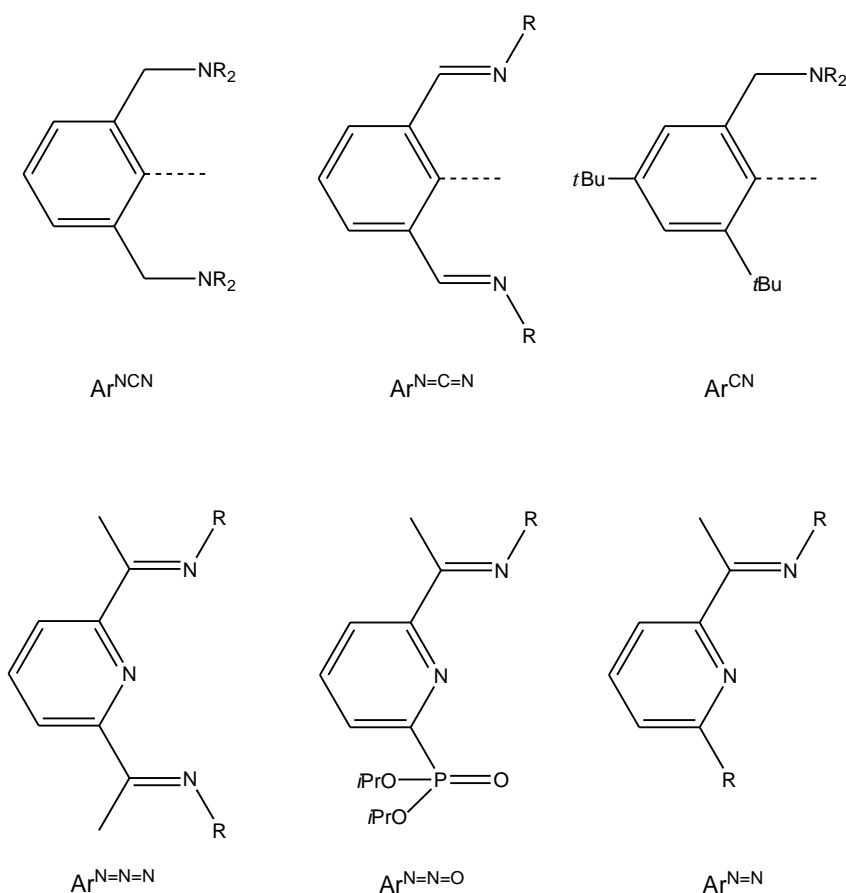


Figure 1. Examples of chelating ligands studied in group 14 elements

References:

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